NON LÍNEAR DOUBLE DAY, SEVILLA 2004

Low Temperature Reconstructive Transformations Low Temperature Reconstructive Transformations

MD Alba, JM Trillo and M Naranjo

## SILICATE MINERALS



#### Libeau classification

Q <sup>n</sup>	Anion group	Silicate Family	Example	Structure
$Q^0$	(SiO <sub>4</sub> ) <sup>4-</sup>	nesosilicate	Forsterite (Mg <sub>2</sub> SiO <sub>4</sub> ) <b>TEOS:</b> (CH <sub>3</sub> CH <sub>2</sub> ) <sub>4</sub> SiO <sub>4</sub>	<b>X</b>
Q <sup>1</sup>	(Si <sub>2</sub> O <sub>7</sub> ) <sup>6-</sup>	Soro- or disilicate	Lu <sub>2</sub> Si <sub>2</sub> O <sub>7</sub>	X
Q <sup>2</sup>	$(\mathrm{Si}_{n}\mathrm{O}_{2n+n})^{2n-1}$	cyclosilicate	Beryl: Al <sub>2</sub> Be <sub>3</sub> Si <sub>6</sub> O <sub>18</sub>	<b>Å</b>
	(SiO <sub>3</sub> ) <sup>2-</sup>	inosilicate	Enstatite: Mg <sub>2</sub> Si <sub>2</sub> O <sub>6</sub>	5
Q <sup>3</sup>	(Si <sub>4</sub> O <sub>10</sub> ) <sup>4-</sup>	phyllosilicate	Saponite: $Na_x[Si_{8-x}Al_x][Mg_6]O_{20}(OH)_4$ Muscovite: $K_2[Si_4Al_2][Al_4]O_{20}(OH)_4$	
Q <sup>4</sup>	SiO <sub>2</sub>	tectosilicate	Silica gel: SiO <sub>2</sub>	

## SILICATE MINERALS: Phyllosilicates

#### **BUILDING BLOCKS**

Tetrahedral Layer (Si<sub>6+x</sub>Al<sub>2-x</sub>O<sub>20</sub>)<sup>10-</sup>





## SILICATE MINERALS: Phyllosilicates

Layer	Layer Charge (X)	Group	Species	
Туре			Dioctahedral member	Trioctahedral member
1:1	x=0	Kaolinite- serpentine	Kaolinite Halloysite	Chrisotile
	x=0	Pyrophyllite-Talc	Pyrophyllite	Talc
	0.2?x ?0.6	Smectite	Montmorillonite Beidellite nontronite	Saponite Hectorite
2:1	0.6 ?x ?0.9	Vermiculite	Dioctahedral vermiculite	Trioctahedral vermiculite
	x=1	Mica	Muscovite Paragonite	Phlogopite Biotite
	x=2	Brittle mica	Margarite	Clintonite Mica-4

### **SOLID-SOLID TRANFORMATIONS**

\* **Displacive Transformation:** It involves only small adjustements to the crystal structure. Generally, no bonds are broken, but the angles between the atoms may change slightly.

\* **Reconstructive Transformation:** It involves extensive rearrangement of the crystal structure and requires breaking of chemical bonds and reassembling the atoms into a different crystal structure.

α-quartz Tridymite **B**-quartz

\* Order-Disorder Transformation: It involves the state of order or disorder in a crystal structure. <sup>29</sup>Si MAS NMR



Day, Sevilla 2004.

Sample	Site	δ (ppm)	/
NAT001†	Si2(3AI)	-87.6	0.668
	Si1(2AI)	-95.2	0.332
NAT002†	Si2(4AI)	-83.6	.082
	Si2(3AI)	-87.6	.512
	Si1(3AI)	-90.0	.051
	Si2(2AI)	-92.1	.044
	Si1(2AI)	-95.1	.260
	Si2(1AJ)	-97.6	.052



## **RECONSTRUCTIVE TRANFORMATION:** Phyllosilicates

Laver	Layer Charge (X)	Group	Species	
Туре			Dioctahedral member	Trioctahedral member
1:1	x=0	Kaolinite- serpentine	Kaolinite Halloysite	Chrisotile
	x=0	Pyrophyllite-Talc	Pyrophyllite	Talc
	0.2?x ?0.6	Smectite	Montmorillonite Beidellite nontronite	Saponite Hectorite
2:1	0.6 ?x ?0.9	Vermiculite	Dioctahedral vermiculite	Trioctahedral vermiculite
	x=1	Mica	Muscovite Paragonite	Phlogopite Biotite
	x=2	Brittle mica	Margarite	Clintonite Mica-4

# **RECONSTRUCTIVE TRANFORMATION:** Smectite





## **RECONSTRUCTIVE TRANFORMATION:** Phyllosilicates

Laver	Layer Charge (X)	Group	Species	
Туре			Dioctahedral member	Trioctahedral member
1:1	x=0	Kaolinite- serpentine	Kaolinite Halloysite	Chrisotile
	x=0	Pyrophyllite-Talc	Pyrophyllite	Talc
	0.2?x ?0.6	Smectite	Montmorillonite Beidellite nontronite	Saponite Hectorite
2:1	0.6 ?x ?0.9	Vermiculite	Dioctahedral vermiculite	Trioctahedral vermiculite
	x=1	Mica	Muscovite Paragonite	Phlogopite Biotite
	x=2	Brittle mica	Margarite	Clintonite Mica-4

## **RECONSTRUCTIVE TRANFORMATION: Muscovite**

